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**Astronomy  
&  
Astrophysics**

# Transmission spectral properties of clouds for hot Jupiter exoplanets

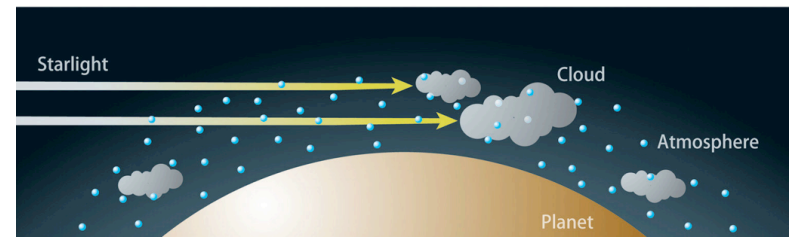
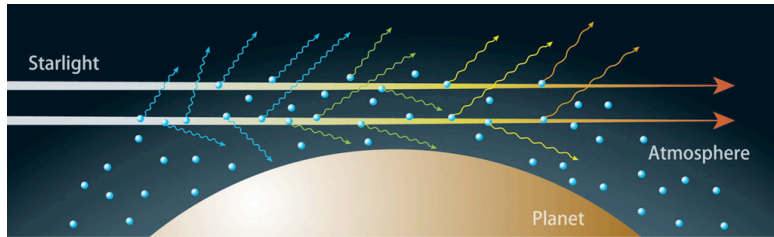
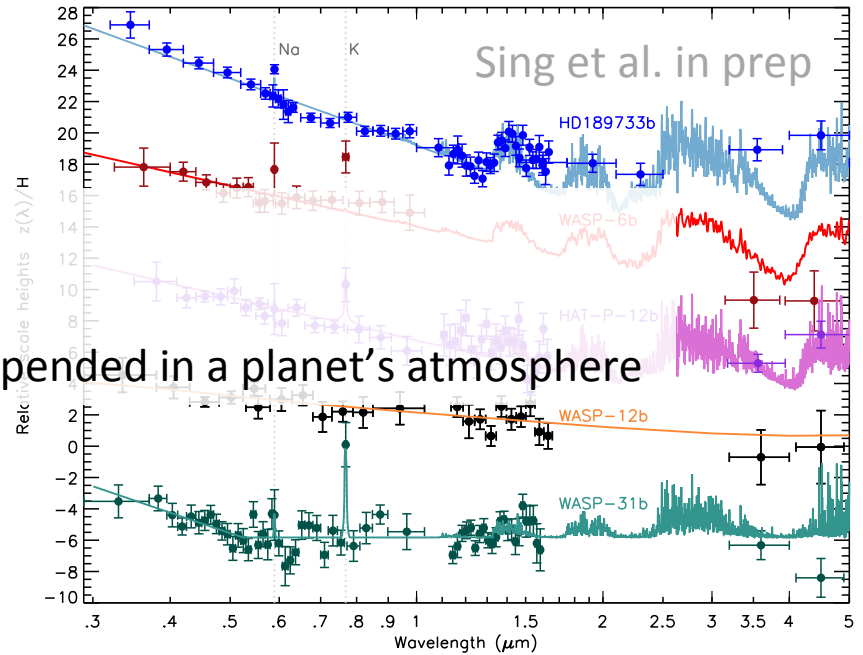
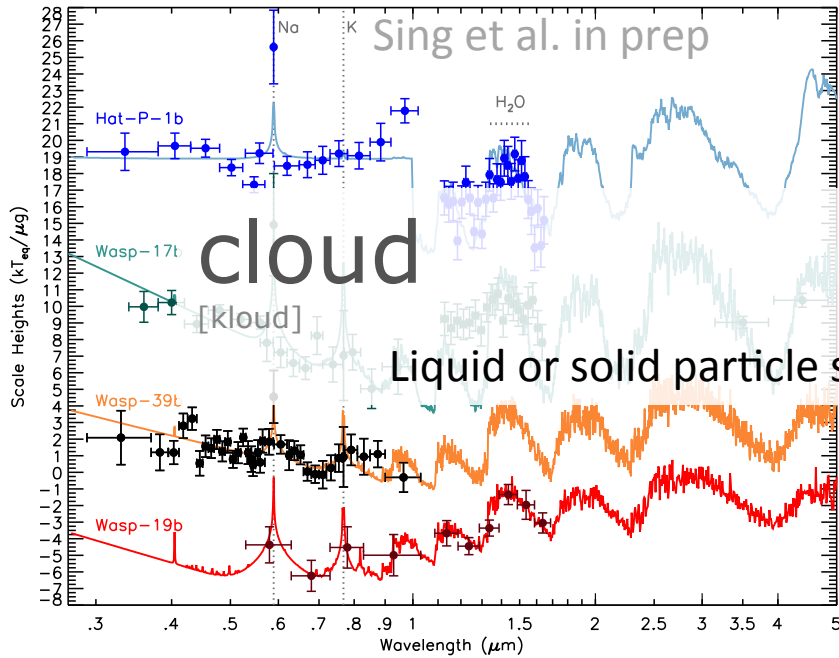
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UNIVERSITY OF  
**EXETER**

# Observations of exoplanet atmospheres



Clear

Wakeford et al. 2013  
Line et al. 2013  
Nikolov et al. 2014  
Huitson et al. 2013

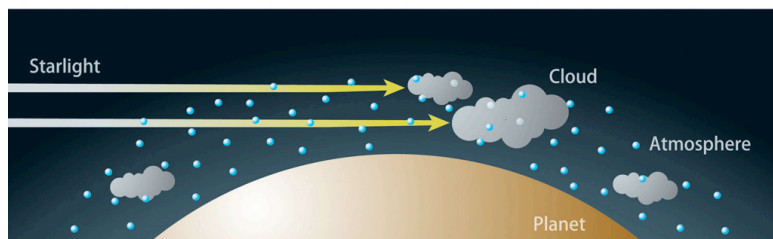
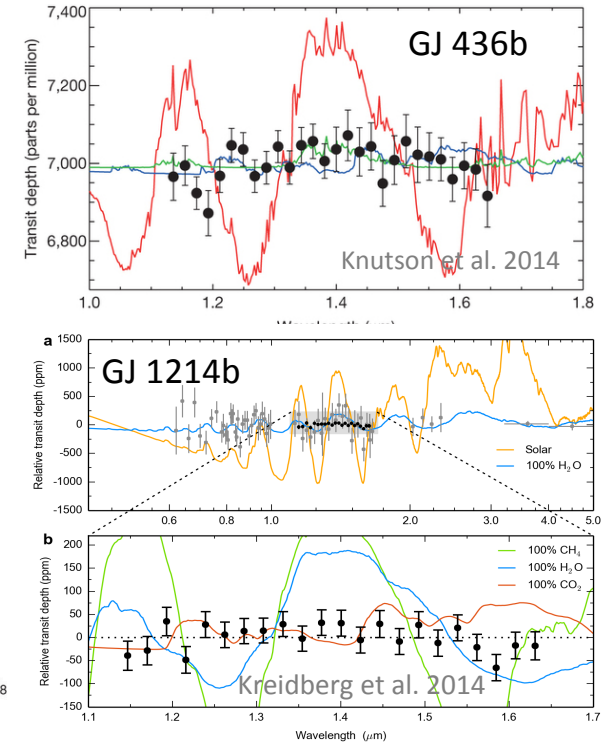
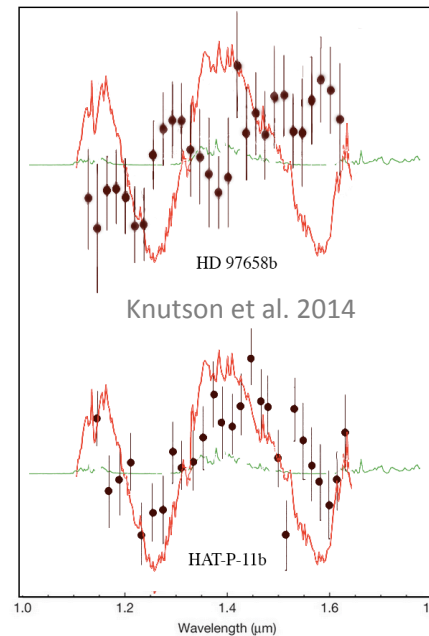
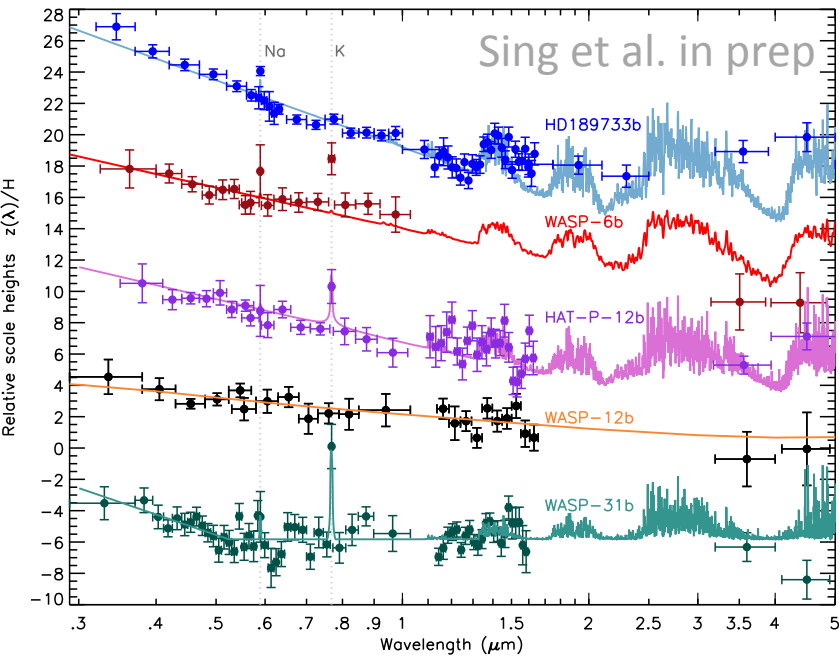
Pont et al. 2008  
McCullough et al. 2014  
Nikolov et al. 2015  
Sing et al. 2013

Sing et al. 2015  
Ballester et al. in prep  
Sing et al. in prep  
Fortney et al. (2010)

Cloudy/Hazy

# Clouds in exoplanet atmospheres?

Clouds obscure absorption features across the optical and often near-IR



Cloudy/Hazy

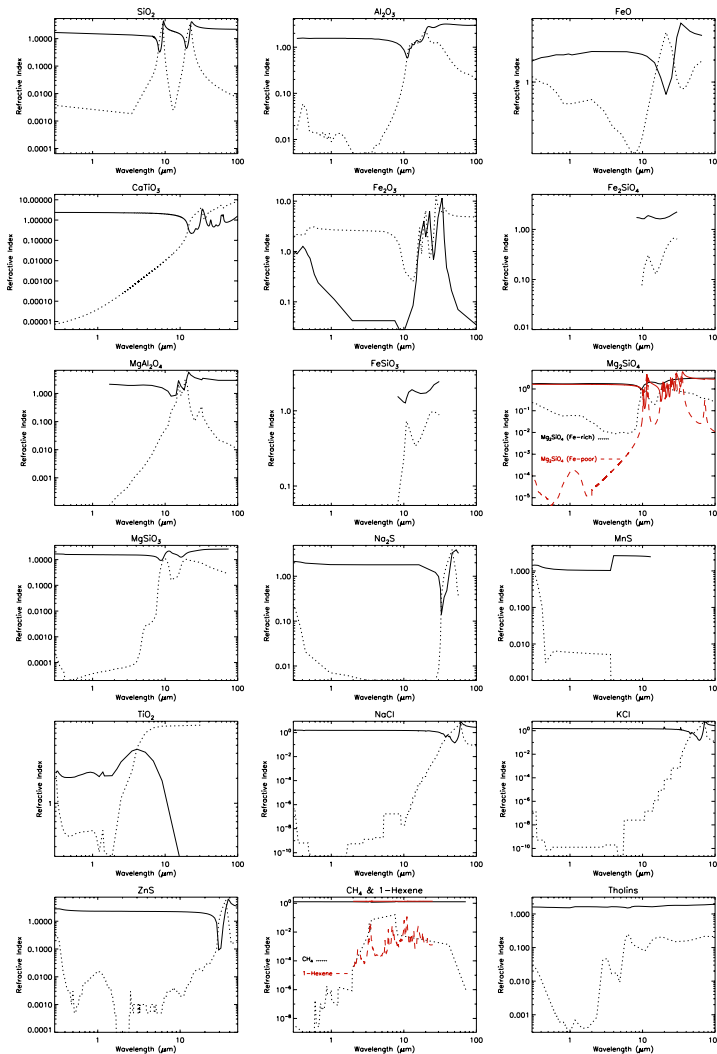
ISM and Brown dwarf studies have shown that it is possible to differentiate between spectral absorption of different species







# Optical Properties of condensates



Wakeford & Sing (2015)

**Table 1:** Table of references for n and k index for a number of condensates expected to form clouds in the upper atmosphere of hot Jupiters.

Condensate	Reference n, k index	$\lambda$ Range ( $\mu\text{m}$ )	Condensation Temperature <sup>+</sup> (K)	Molecular Weight
SiO <sub>2</sub>	Palik (1998)	0.04 - 11	1725	60.08
	Andersen et al. (2006)	7 - 28	-	-
	M. Meinecke (2005)*	6.6 - 10000	-	-
Al <sub>2</sub> O <sub>3</sub>	Koike et al. (1995)	0.3 - 150	1677 <sup>1</sup>	101.96
	Begemann et al. (1995)	10 - 100	1650 <sup>4</sup>	71.79
FeO	Andersen et al. (2006)	15 - 40	-	-
CaTiO <sub>3</sub>	Posch et al. (2003)	2 - 155	1582 <sup>1</sup>	135.94
Fe <sub>2</sub> O <sub>3</sub>	M. Meinecke (2005)*	0.1 - 987	1566	159.68
Fe <sub>2</sub> SiO <sub>4</sub>	Day (1981)	8.2 - 35	1443 <sup>4</sup>	203.77
MgAl <sub>2</sub> O <sub>4</sub>	M. Meinecke (2005)*	1.6 - 270	1397 <sup>1</sup>	142.26
FeSiO <sub>3</sub>	Day (1981)	8.2 - 35	1366 <sup>4</sup>	131.92
Mg <sub>2</sub> SiO <sub>4</sub> (Fe-rich)	Henning et al. (2005)	0.2 - 445	1354 <sup>1</sup>	140.63
Mg <sub>2</sub> SiO <sub>4</sub> (Fe-poor)	Zeidler et al. (2011)	0.19 - 800	1354 <sup>1</sup>	140.63
MgSiO <sub>3</sub>	Egan & Hilgeman (1975)	0.1 - 0.4	1316 <sup>1</sup>	100.33
	Dorschner et al. (1995)	0.5 - 80	-	-
Na <sub>2</sub> S	Morley et al. (2012)	0.03 - 73	1176	78.04
MnS	Huffman & Wild (1967)	0.1 - 3	1139 <sup>2</sup>	87.00
TiO <sub>2</sub>	Kangaroo (2010a)	0.3 - 1.2	1125 <sup>2</sup>	79.86
	Kangaroo (2010b)	1.3 - 30	-	-
NaCl	Palik (1998)	0.04 - 1000	825 <sup>3</sup>	58.44
KCl	Palik (1998)	0.02 - 200	740 <sup>3</sup>	74.55
ZnS	Querry (1987)	0.2 - 167	700 <sup>5</sup>	97.45
CH <sub>4</sub>	Martonchik & Orton (1994)	0.02 - 72	~80	16.04
C <sub>6</sub> H <sub>12</sub>	Anderson (2000)	2.0 - 25	68	84.1
Titan Tholins	Khare et al. (1984)	0.01 - 0.2	≤90	~50.0
	-	1.1 - 1000	-	-
	Ramirez et al. (2002)	0.2 - 1	-	-

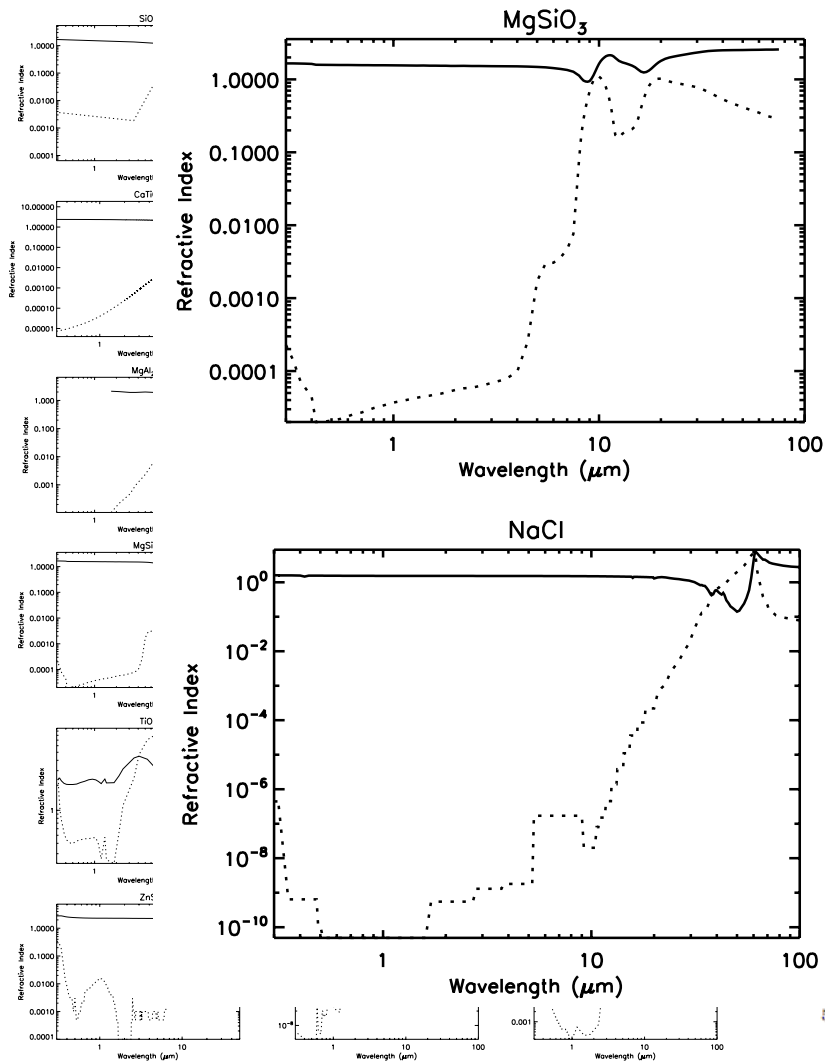
\* <http://www.astro.uni-jena.de/Laboratory/OCDB/oxsul.html>; <sup>+</sup> at 10<sup>-3</sup> bar

<sup>1</sup> Lodders (2003), <sup>2</sup> Grossman (1972), <sup>3</sup> Burrows & Sharp (1999), <sup>4</sup> Ebel & Grossman (2000)

<sup>5</sup> Morley et al. (2012)

Wakeford & Sing (2015)

# Optical properties of condensates

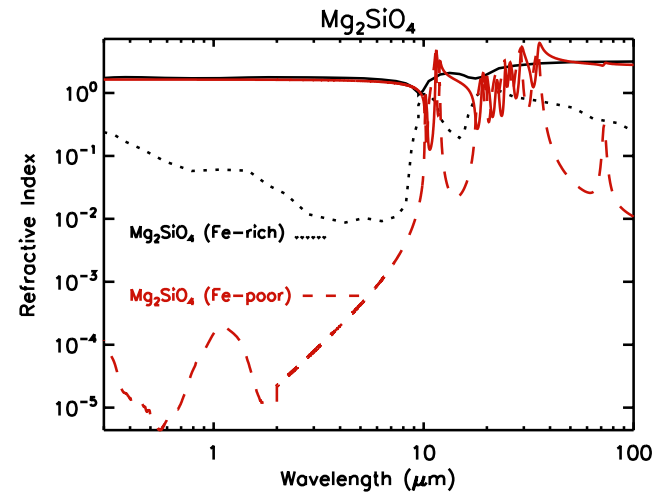


Wakeford & Sing (2015)

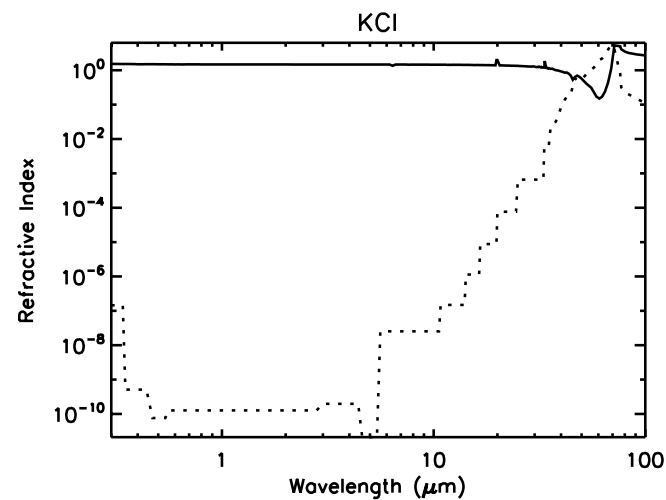
Table 1: 1  
clouds  
Condense

- SiC
- Al<sub>2</sub>O<sub>3</sub>
- Fe
- CaTi
- Fe<sub>2</sub>O<sub>3</sub>
- Fe<sub>2</sub>S
- MgAl
- FeSi<sub>2</sub>
- SiO<sub>2</sub> (Fe-rich)
- SiO<sub>2</sub> (Fe-poor)
- MgS
- Na<sub>2</sub>SO<sub>4</sub>
- MgSO<sub>4</sub>
- Na<sub>2</sub>CO<sub>3</sub>
- KCl
- ZnS
- CF<sub>4</sub>
- C<sub>6</sub>H<sub>6</sub>
- Titanium

http://www  
codders  
Morley et



Hawkins et al. (2005) 0.9 - 445 12541

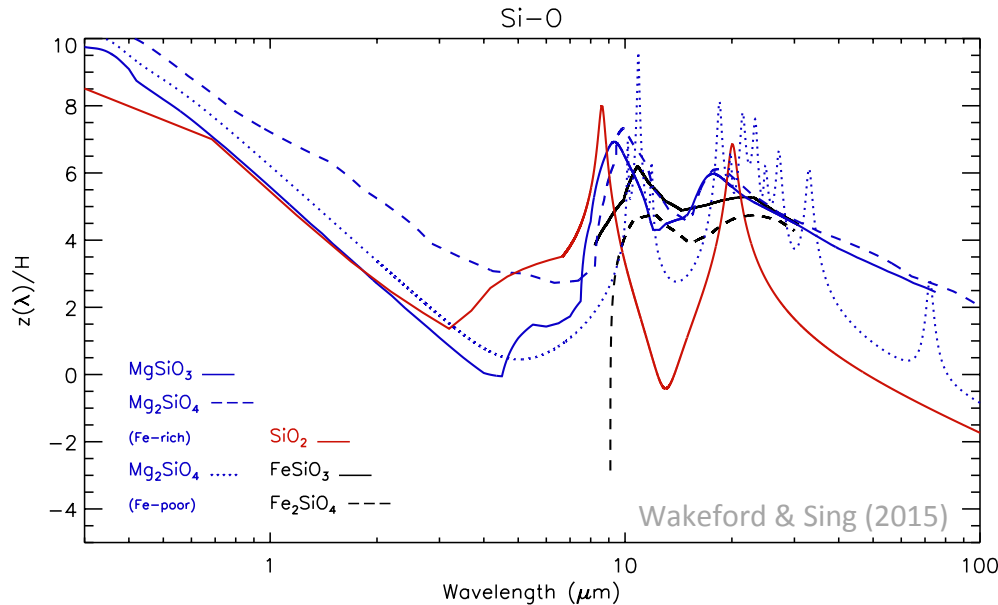


expected to

Ion	Molecular Weight
Si <sup>4+</sup>	60.08
Al <sup>3+</sup>	-
Fe <sup>2+</sup>	101.96
Fe <sup>3+</sup>	71.79
Ca <sup>2+</sup>	-
Ti <sup>4+</sup>	135.94
Fe <sup>2+</sup>	159.68
Fe <sup>3+</sup>	203.77
Mg <sup>2+</sup>	142.26
Fe <sup>2+</sup>	131.92
Si <sup>4+</sup>	140.63
Si <sup>4+</sup>	140.63
K <sup>+</sup>	100.33
Na <sup>+</sup>	-
Na <sup>+</sup>	78.04
Mg <sup>2+</sup>	87.00
Ti <sup>4+</sup>	79.86
Al <sup>3+</sup>	-
Al <sup>3+</sup>	58.44
Al <sup>3+</sup>	74.55
Al <sup>3+</sup>	97.45
Al <sup>3+</sup>	16.04
Al <sup>3+</sup>	84.1
Al <sup>3+</sup>	~50.0
Al <sup>3+</sup>	-
Al <sup>3+</sup>	-

rossman (2000)

# Vibrational modes



Absorption features are dominated by vibrational modes of the major di-atomic bond in the condensate.

**Table 4.3.1:** Table of vibrational modes for the major diatomic bond species in the different cloud condensates considered in this paper.

Major Bond	Reduced Mass, $\mu_M$ (g)	Vibrational Frequency, $\nu$ ( $\text{cm}^{-1}$ )	Wavelength, $\lambda$ ( $\mu\text{m}$ )
Si - O	10.192	1110 - 830 <sup>a</sup>	9 - 12
Al - O	10.043	1100 - 350 <sup>c</sup>	9 - 28.7
Fe - O	12.436	790 <sup>b</sup>	12.5
Ti - O	11.99	850 - 150 <sup>d</sup>	11.7 - 66
MnS	20.247	295-220 <sup>e</sup>	33.8-45.4
ZnS	21.51	464 <sup>f</sup>	21.5
NaCl	13.95	366 <sup>g</sup>	27.32
KCl	18.60	281 <sup>g</sup>	35.5
C - H	0.923	3032 <sup>a</sup>	3.3

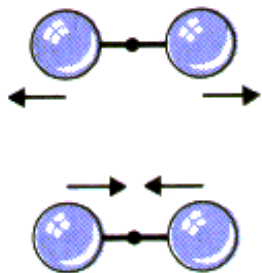
<sup>a</sup>Glassgold & Graham (2008); <sup>b</sup>Lehnert et al. (2002)

<sup>c</sup>Saniger (1995); <sup>d</sup>Gillet et al. (1993)

<sup>e</sup>Batsanov & Derbeneva (1969); <sup>f</sup>Kröger & Meyer (1954)

<sup>g</sup>Rice & Klemperer (2004)

Wakeford & Sing (2015)

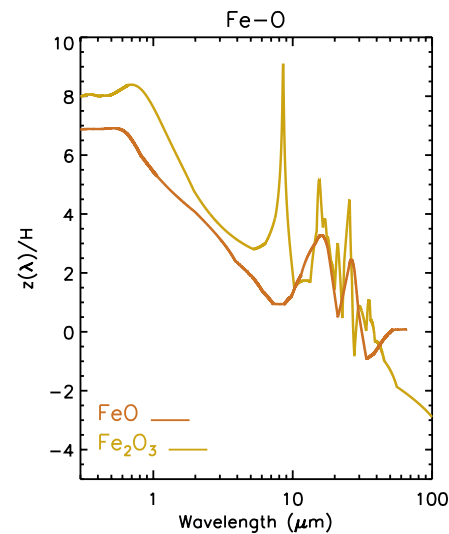
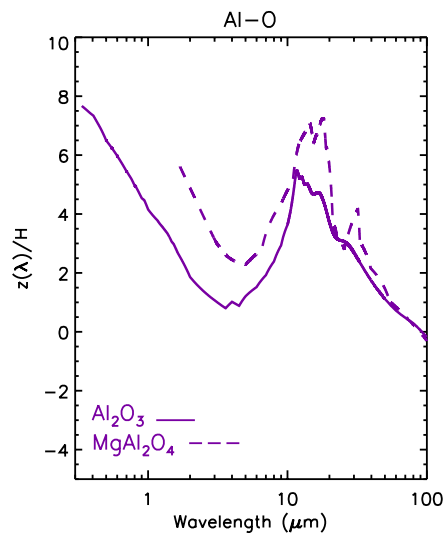
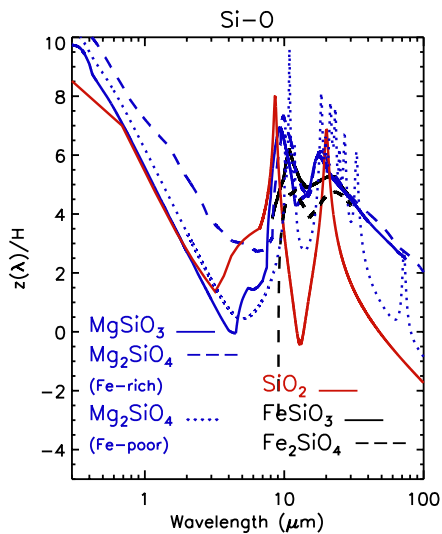


Vibrational motion along the bond between two atoms

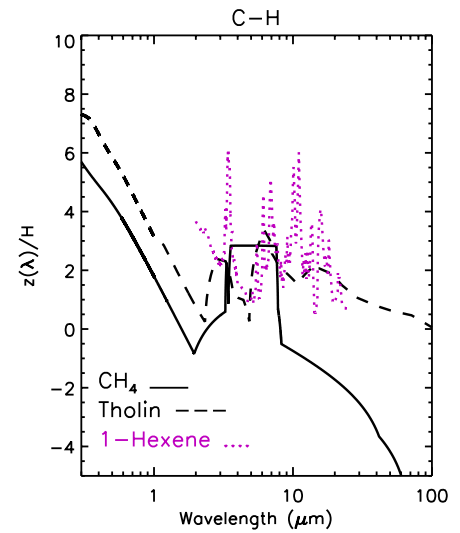
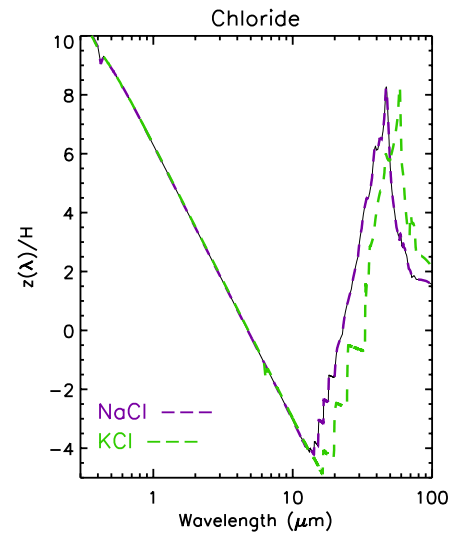
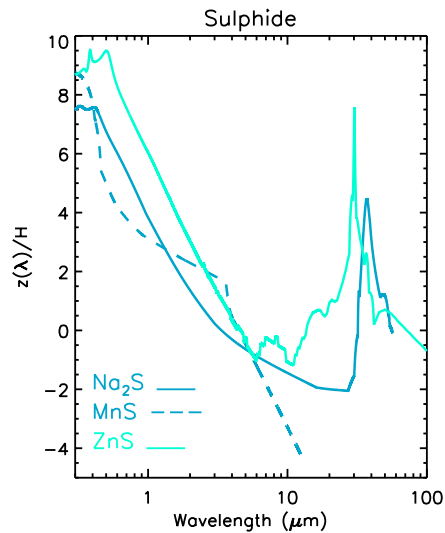
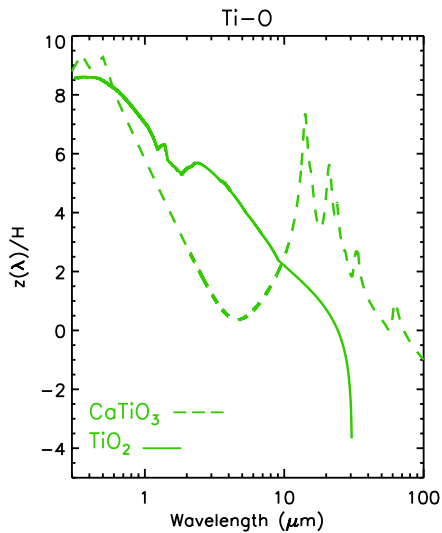
$$\nu = \frac{1}{2\pi c} \sqrt{\frac{K}{\mu_M}}$$



# Transmission Spectra of cloud condensates

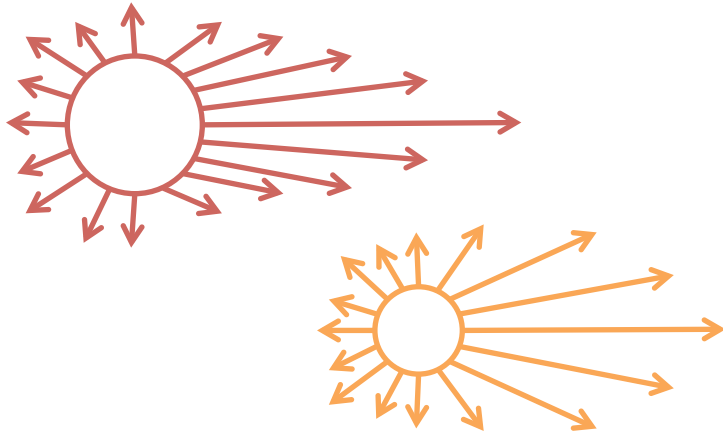


As a result on observational distinction can be made between species generated photochemically or through condensation.



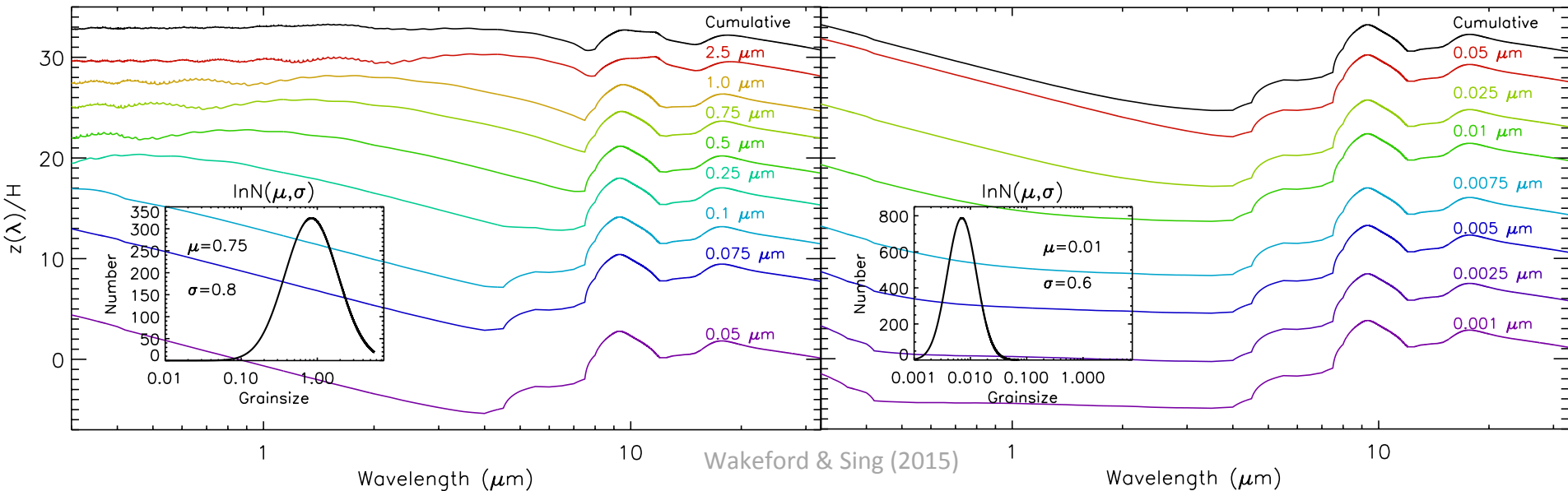
Wakeford & Sing (2015)

# Particle size distributions

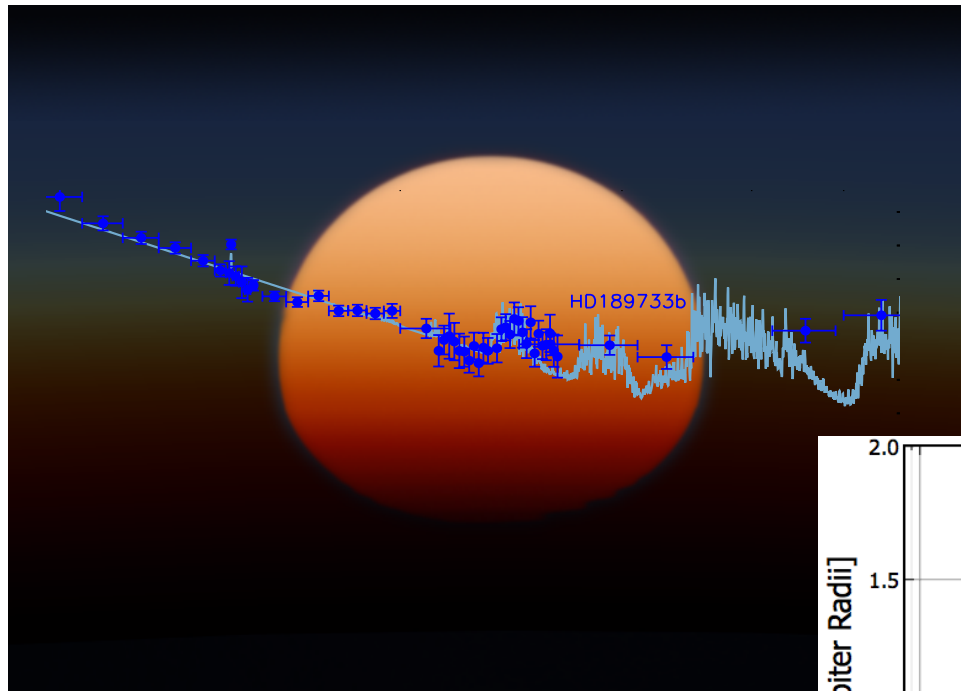


Rayleigh cross-section  $\propto a^6$

This results in the transmission spectrum largely dependent on the largest grain size in the particle distribution.

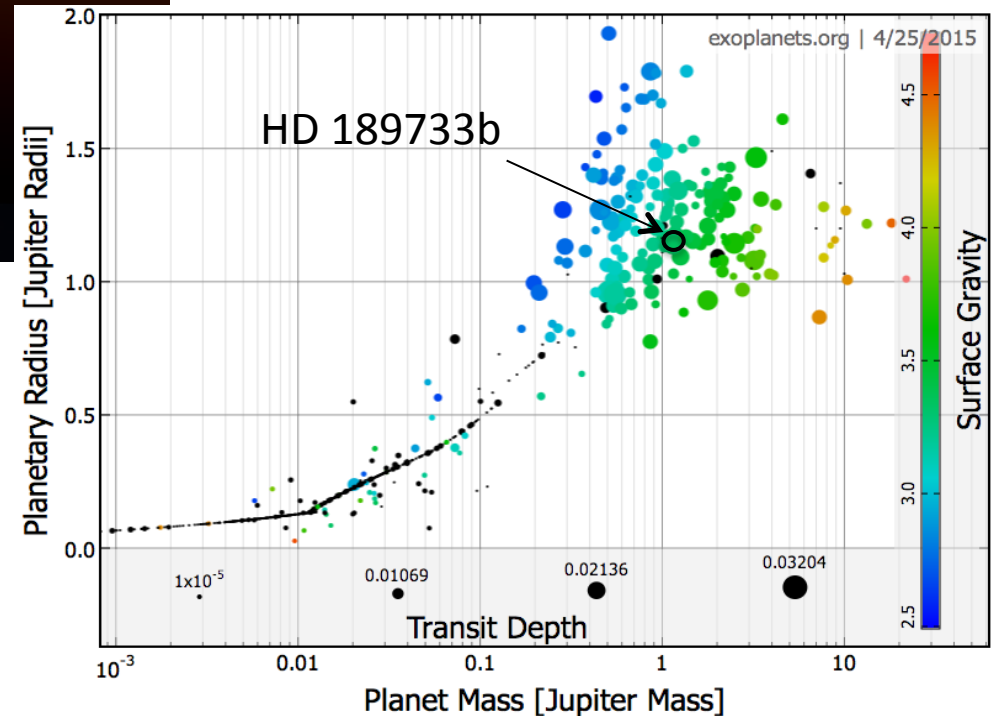


# Hot Jupiter transmission spectra



We use HD 189733b as an example hot Jupiter.

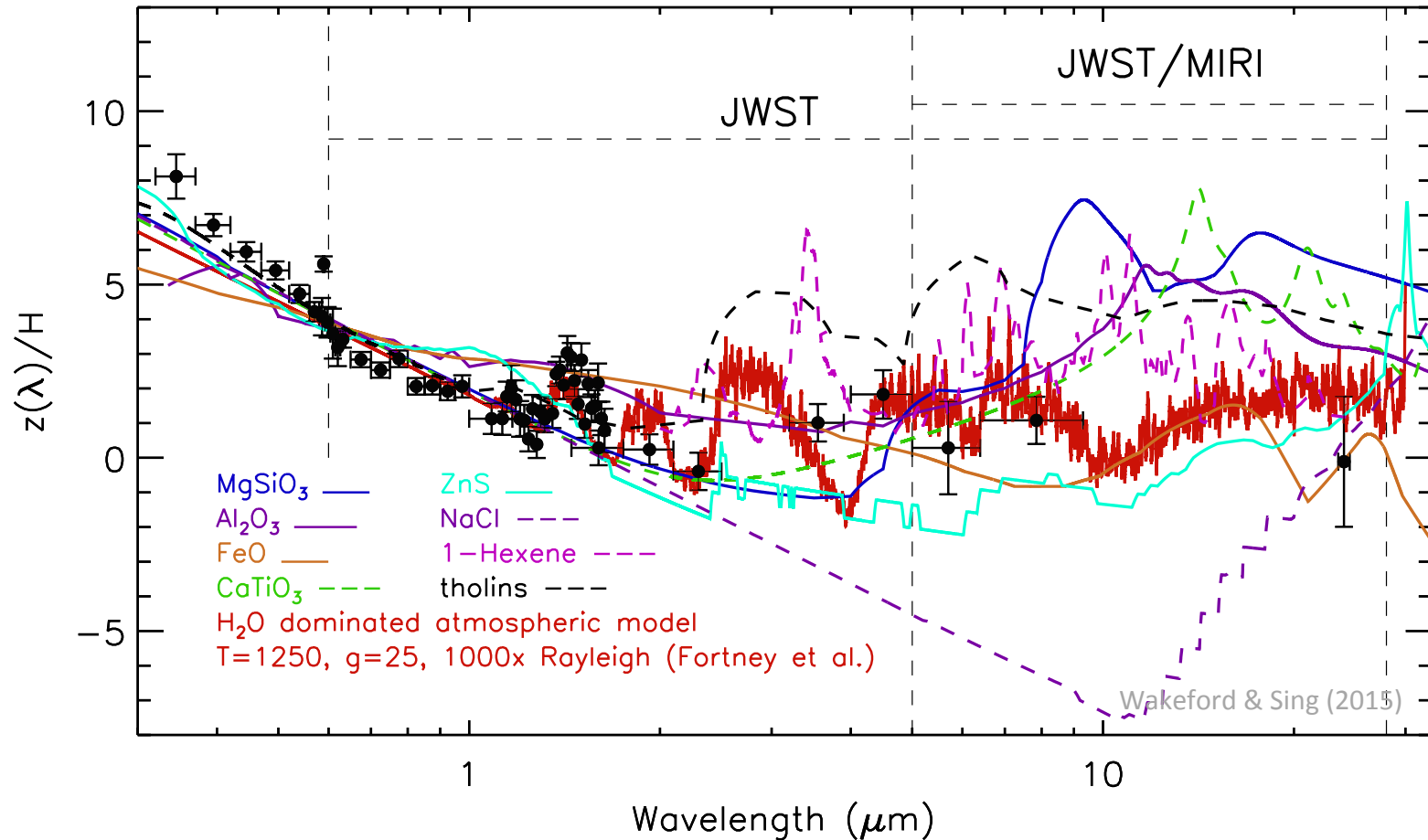
HD 189733b's atmosphere shows strong Rayleigh scattering below  $1\mu\text{m}$  and little indication of the expected alkali metal lines. (Pont et al. 2008)





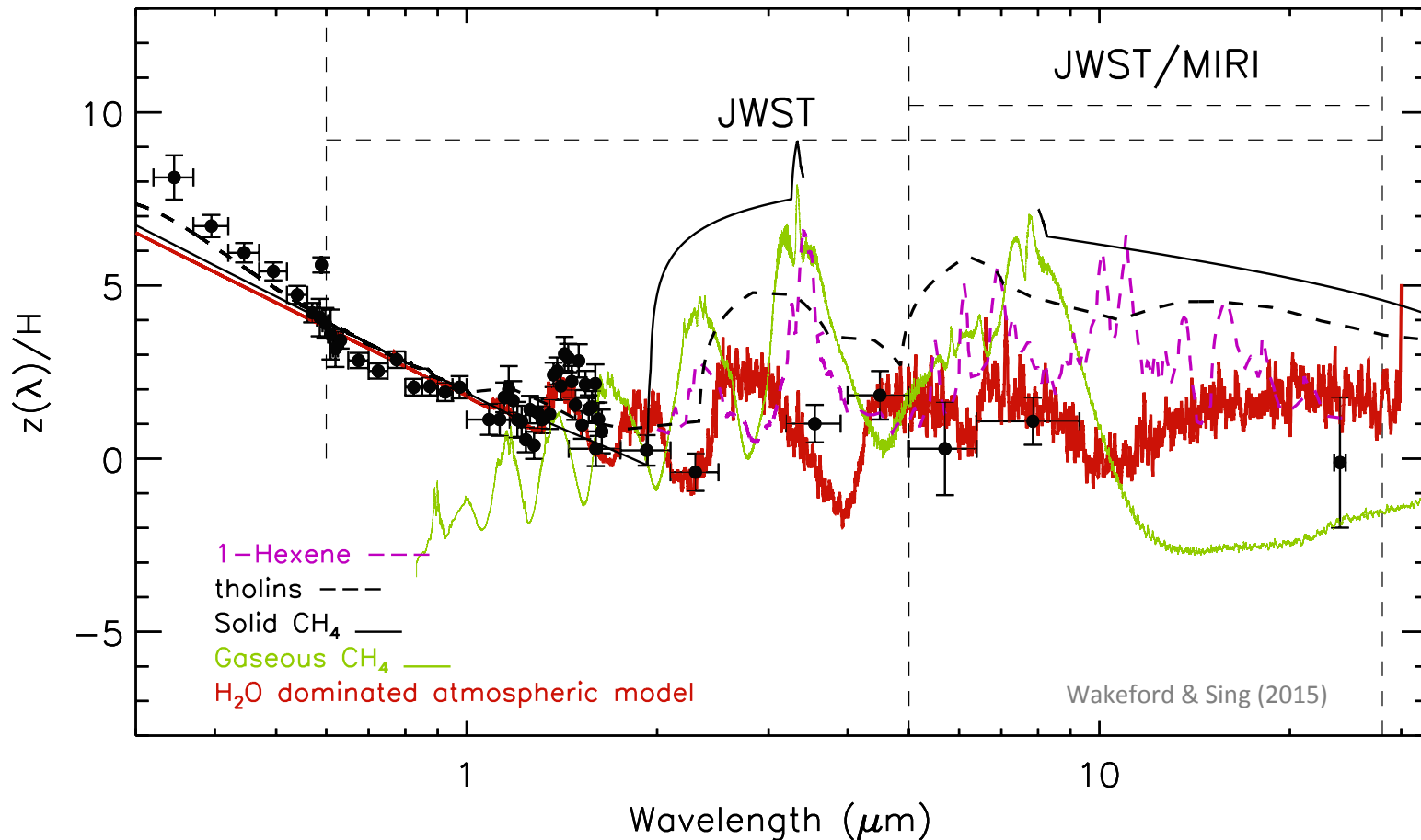
# Condensate transmission spectra

Rayleigh scattering in the optical can be used as a diagnostic for particle size and condensates to be observed in the IR.



# C-H gaseous vs. condensate

Some notable condensate features may be used to differentiate between gaseous and solid particle spectra



Yurchenko & Tennyson (2014)

Amundsen et al. (2014)

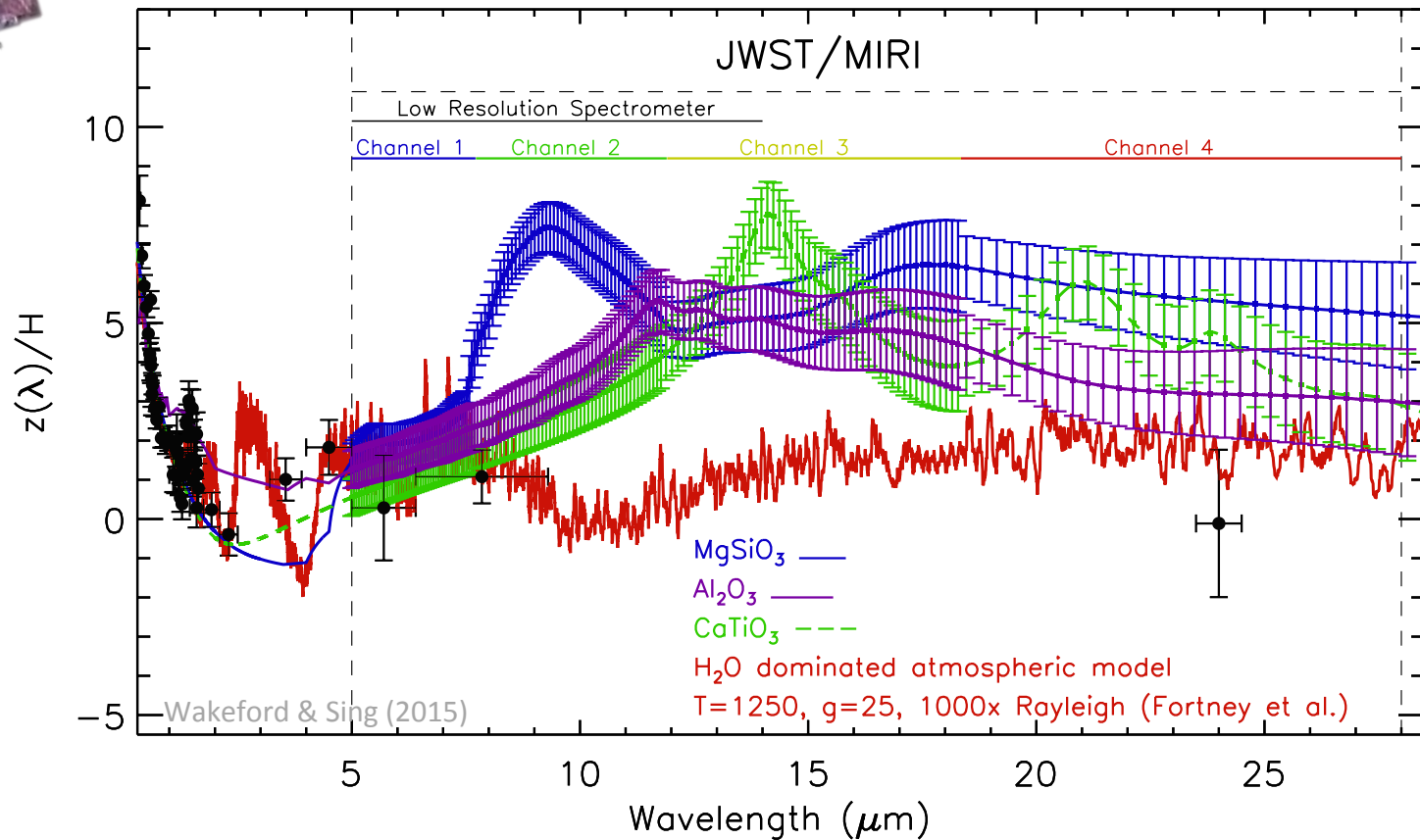


Sharp & Burrows (2007)

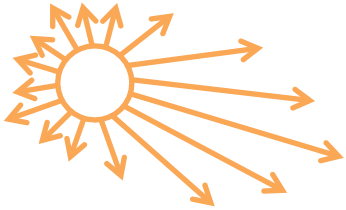
Rothman et al. (2009)

# James Webb Space Telescope

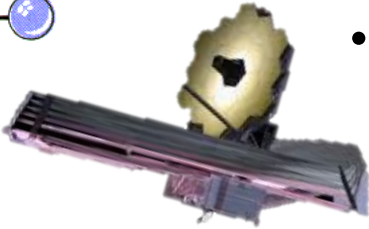
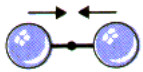
JWST MIRI will have 50x sensitivity and 7x angular resolution of Spitzer.



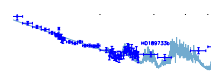
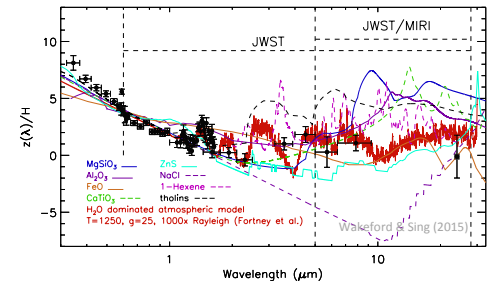




- Condensate transmission spectra can be approximated using the largest particle size in a distribution.
- Absorption features are dominated by vibrational modes of the major di-atomic bond in the condensate.
- As a result an observational distinction can be made between species generated photochemically or through condensation.



- Rayleigh scattering in the optical can be used as a diagnostic for particle size and condensates to be observed in the IR.
- JWST MIRI will have 50x sensitivity and 7x angular resolution of Spitzer.



# Cloudy with a chance of H<sub>2</sub>O

by  
Hannah Wakeford

## What's Next?

NPP Fellow  
working with Avi Mandell

